



## Shri Vaishnav Vidyapeeth Vishwavidyalaya, Indore

### ML307 ENVIRONMENTAL MANAGEMENT AND SUSTAINABILITY

SUBJECT CODE	CATEGORY	SUBJECT NAME	TEACHING & EVALUATION SCHEME								
			THEORY			PRACTICAL		L	T	P	CREDITS
			END SEM University Exam	Two Term Exam	Teachers Assessment*	END SEM University Exam	Teachers Assessment*				
ML-307	Compulsory	Environmental Management and Sustainability	60	20	20	0	0	4	0	0	4

**Legends:** L - Lecture; T - Tutorial/Teacher Guided Student Activity; P – Practical; C - Credit;

\***Teacher's Assessment** shall be based upon following components: Quiz/Assignment/ Project/Participation in Class, given that no component shall exceed more than 10 marks.

#### Course Objectives

1. To create awareness towards various environmental problems.
2. To create awareness among students towards issues of sustainable development.
3. To expose students towards environment friendly practices of organizations.
4. To sensitize students to act responsibly towards environment.

#### Examination Scheme

The internal assessment of the students' performance will be done out of 40 Marks. The semester Examination will be worth 60 Marks. The question paper and semester exam will consist of two sections A and B. Section A will carry 36 Marks and consist of five questions, out of which student will be required to attempt any three questions. Section B will comprise of one or more cases / problems worth 24 marks.

#### Course Outcomes

1. The course will give students an overview of various environmental concerns and practical challenges in environmental management and sustainability.
2. Emphasis is given to make students practice environment friendly behavior in day-to-day activities.



# **Shri Vaishnav Vidyapeeth Vishwavidyalaya, Indore**

## **COURSE CONTENT**

### **Unit I: Introduction to Environment Pollution and Control**

1. Pollution and its types (Air, Water, and Soil): Causes, Effects and Control measures
2. Municipal Solid Waste: Definition, Composition, Effects
3. Electronic Waste: Definition, Composition, Effects
4. Plastic Pollution: Causes, Effects and Control Measures

### **Unit II: Climate Change and Environmental Challenges**

1. Global Warming and Green House Effect
2. Depletion of the Ozone Layer
3. Acid Rain
4. Nuclear Hazards

### **Unit III: Environmental Management and Sustainable Development**

1. Environmental Management and Sustainable Development: An overview
2. Sustainable Development Goals (17 SDGs)
3. Significance of Sustainable Development
4. Environment Friendly Practices At Workplace and Home (Three Rs' of Waste Management, Water Conservation, Energy Conservation)

### **Unit IV: Environmental Acts**

1. The Water (Prevention and Control of Pollution) Act, 1974: Objectives, Definition of Pollution under this act, Powers and Functions of Boards
2. The Air (Prevention and Control of Pollution) Act, 1981: Objectives, Definition of Pollution under this act, Powers and Functions of Boards
3. The Environment (Protection) Act, 1986: Objectives, Definition of important terms used in this Act, Details about the act.
4. Environmental Impact Assessment: Concept and Benefits

### **Unit V: Role of Individuals, Corporate and Society**

1. Environmental Values
2. Positive and Adverse Impact of Technological Developments on Society and Environment
3. Role of an individual/ Corporate/ Society in environmental conservation
4. Case Studies: The Bhopal Gas Tragedy, New Delhi's Air Pollution, Arsenic Pollution in Ground Water (West Bengal), Narmada Valley Project, Cauvery Water Dispute, Fukushima Daiichi Disaster (Japan), Ozone Hole over Antarctica, Ganga Pollution, Deterioration of Taj Mahal, Uttarakhand flash floods



# Shri Vaishnav Vidyapeeth Vishwavidyalaya, Indore

## Name of the Program: B. Sc. (Mathematics Honours)

SUBJECT CODE	Category	SUBJECT NAME	TEACHING & EVALUATION SCHEME								
			THEORY			PRACTICAL		Th	T	P	CREDITS
			END SEM	MST	Q/A	END SEM	Q/A				
BSMHMA 302	DC	Algebra-III (Linear Algebra)	60	20	20	-	-	4	1	-	5

### Course Objective

*To introduce the students with the Linear Algebra.*

### Course Outcomes

*After the successful completion of this course students will be able to*

- 1. understand the application of Matrices*
- 2. know the fundamentals of the Vector Space.*
- 3. know the fundamentals of the Inner Product Space.*

### Course Content:

**Unit I :** Vector spaces: subspaces and quotient spaces; homomorphism and isomorphism theorems

**Unit II:** Linearly Dependent and Independent of Vectors, Bases and dimension of a vector space.

**Unit III :** Linear transformations and their matrices; row and column spaces of a matrix; Rank- Nullity theorem; system of linear equations.

**Unit IV :** Eigen-values and eigenvectors; Caley-Hamilton Theorem.

**Unit V :** Binary and Quadratic forms and reduction into Canonical Forms. Inner product spaces; Orthogonal Basis; Gram-Schmidt Orthogonalization Process.



# **Shri Vaishnav Vidyapeeth Vishwavidyalaya, Indore**

## **Name of the Program: B. Sc. (Mathematics Honours)**

### **Reference Books :**

1. M. Artin: Algebra, Pearson.
2. S. D. Dummit and M. R. Foote: Abstract Algebra, Wiley.
3. I. N. Herstein: Topics in Algebra, Wiley.
6. K. Hoffman and R. Kunze: Linear Algebra, Prentice Hall of India.
7. F. E. Hohn: Elementary Matrix Algebra, Dover Publications.
8. P. R. Halmos: Finite Dimensional Vector Spaces, Springer New York.
9. S. Lang, Introduction to Linear Algebra, Wellesley - Cambridge Press.
10. S. Kumaresan, Linear Algebra- A Geometric Approach, Prentice Hall of India.



# Shri Vaishnav Vidyapeeth Vishwavidyalaya, Indore

## Name of the Program: B. Sc. (Mathematics Honours)

SUBJECT CODE	Category	SUBJECT NAME	TEACHING & EVALUATION SCHEME								
			THEORY			PRACTICAL		Th	T	P	CREDITS
			END SEM	MST	Q/A	END SEM	Q/A				
BSMHMA 303	DC	Metric Topology	60	20	20	-	-	4	1	-	5

### Course Objective

*To introduce the students with the Metric Topology.*

### Course Outcomes

*After the successful completion of this course students will be able to understand and apply the basics of the Metric Spaces.*

### Course Content:

#### Unit I:

Elements of metric space theory, sequences and Cauchy sequences and the notion of completeness.

#### Unit II:

Construction of real numbers, elementary topological notions for metric spaces i.e. open sets, closed sets, compact sets, connectedness.

#### Unit III:

Continuous and uniformly continuous functions on a metric space. The Bolzano - Weierstrass theorem, supremum and infimum on compact sets.

#### Unit IV:

Separability. Completeness. The Baire Category Theorem.  $R_n$  as a metric space.

#### Unit V:

Contractions on Metric Spaces and their examples, Banach Contraction Principle,  $C(X)$  as a complete metric space, Picard's Existence Theorem.



# **Shri Vaishnav Vidyapeeth Vishwavidyalaya, Indore**

## **Name of the Program: B. Sc. (Mathematics Honours)**

### **Reference Books :**

1. S. Kumaresan, Topology of Metric Spaces, Narosa Publishing House.
2. W. Rudin: Principles of Mathematical Analysis, Mac Graw Hill Education.
3. Tom Apostol: Mathematical Analysis, Pearson.
4. Tom Apostol: Calculus I and II, Pearson.
5. Terence Tao : Analysis I, II, Springer Singapore.
6. W. Rudin: Real and Complex Analysis, Mac Graw Hill Education.



# Shri Vaishnav Vidyapeeth Vishwavidyalaya, Indore

## Name of the Program: B. Sc. (Mathematics Honours)

SUBJECT CODE	Category	SUBJECT NAME	TEACHING & EVALUATION SCHEME								
			THEORY			PRACTICAL		Th	T	P	CREDITS
			END SEM	MST	Q/A	END SEM	Q/A				
BSMHMA 304	DC	Complex Analysis	60	20	20	-	-	3	1	-	4

### Course Objective

*To introduce the students with the Calculus of the Complex Variables.*

### Course Outcomes

*After the successful completion of this course students will be able to understand and apply the basics of the Calculus of the Complex Variables.*

### Course Content:

#### Unit I:

Limit, Continuity and Differentiability of Functions of Complex Variables, Analytic functions, Cauchy-Riemann Equations, Polar form of Cauchy-Riemann Equations, Harmonic functions, Method of constructing an Analytical Function.

#### Unit II:

Complex Line Integrals, Cauchy's Theorem, Extension of Cauchy Theorem, Cauchy's Integral Formula.

#### Unit III:

Liouville's theorem, Zeros of analytic functions, The fundamental theorem of algebra. Singularities of an analytic function, Types of singularities, Power Series, Expansions of Analytic Functions as Power series (Taylor and Laurent Series).

#### Unit IV:

Poles and Residues, Cauchy's Residue theorem, The Argument Principle, Evaluation of Real Definite Integrals.

#### Unit V:

The Maximum Modulus Principle, Schwarz's Lemma, Cross Ratio, Translation, Magnification, Rotation, Inversion, Reflection and Mobius (Bilinear) Transformations, Fixed point, Critical Point, Normal form of Bilinear Transformations. Conformal Transformation.



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## **Name of the Program: B. Sc. (Mathematics Honours)**

### **Reference Books:**

1. Elias M. Stein, Rami Shakarchi: Complex Analysis, Princeton University Press.
2. Lars Ahlfors: Complex Analysis, Mc-Graw Hill.
3. T. W. Gamelin: Complex Analysis, Springer.
4. J.B.Conway: Functions of One Complex Variable, Springer-Verlag New York.
5. S. Ponnusamy, Foundations Of Complex Analysis, Narosa Publishing House.





**Shri Vaishnav Vidyapeeth Vishwavidyalaya, Indore**  
**Name of the Program: B. Sc. (Mathematics Honours)**

**DEGREE PROGRAM**

**B.Sc III Sem**

SUBJECT CODE	Category	SUBJECT NAME	TEACHING &EVALUATION SCHEME								
			THEORY			PRACTICAL		Th	T	P	CREDITS
			End Sem University Exam	Two Term Exam	Teac hers Assessment *	End Sem Unive rsity Exam	Tea cher s Asse ssm ent*				
BSPH302	DC	Electronics:  Principles and Devices	60	20	20	30	20	3	1	4	6

Legends: L - Lecture; T - Tutorial/Teacher Guided Student Activity; P – Practical; C - Credit;

Q/A – Quiz/Assignment/Attendance, MST MidSem Test.

\*Teacher Assessment shall be based on following components: Quiz/Assignment/Project/Participation in class (Given that no component shall be exceed 10 Marks)

**Course Objectives:-**

1. To develop the comprehensive understanding of laws of physics related to Electronics: Principles and Devices and ability to apply them for laying the foundation for research and development.
2. To work ethically as member as well as leader in a diverse team.

## **Course Outcomes:-**

1. Student will be able to understand and solve the problems related to Electronics: Principles and Devices,
2. Student will be able to determine physical parameter experimentally with optimal usage of resources and complete the assignments in time.

## **BSPH 302- Electronics: Principles and Devices**

### **Unit 1:-**

Classical FE Model, Debye Model, Sommerfeld FE Model, Band Model, Kronig-Penney Model, Effective Mass, Formulation of Energy Bands, Gap in Solids, Motion of  $e^-$  in Metals, Density of States, Fermi Level, Fermi Velocity and Fermi Dirac Distribution of  $e^-$  Inside a Matter.

### **Unit-2**

Semiconductors; Intrinsic-semiconductors, electrons and holes, Fermi Level, Temperature dependence of electron and hole concentrations Doping: impurity states, n and p type semiconductors, conductivity, mobility, Hall Effect, Hall Coefficient. Semiconductor devices: Metal-semiconductor junction, p-n junction, majority and minority carriers,

### **Unit-3**

Zener and tunnel diodes, light emitting diode, solar cell Diode as a circuit element, load line concept, rectification, ripple & factor, Zener diode, voltage stabilization, IC voltage regulation. FETs: Field effect transistors JFET, BJT, MOSFET, Transistors, Characteristics of a transistor in CB, CE and CC mode, h-parameters,

### **Unit-4**

Amplifiers, Small signal amplifiers; General Principle of operation, classification, distortion, RC coupled amplifier, gain frequency response, input and output impedance, multistage amplifiers. Transformer coupled amplifiers, Equivalent circuits at low, medium and high frequencies, emitter follower, low frequency common source and common drain amplifier, Noise in electronic circuits.

### **Unit-5**

Oscillators, Feedback in amplifiers, principle, its effects on amplifiers, characteristics Principle of feedback amplifier, Barkhausen criteria, Hartley, Colpitt and Wein bridge oscillators. Condition for oscillations and frequency derivation - Crystal oscillator - UJT Relaxation oscillator. Monostable, Bi-stable and Astable multivibrators

## **References:**

1. Introduction to Solid State Physics C. Kittel
2. Solid State Physics : R.L, Singhal
3. Micro Electronics J- Millman and A. Grabel
4. Electronic Devices and Circuits : MillmanHalkias
5. Electronic Devices Circuits and Applications : J.D. Ryder
6. Electronic Devices and Circuits: Robert Baylested and Louis Nashelsky

### **List of Experiments (Any Eight)**

1. Find V-I characteristics of PN Junction Diode.
2. To Find V-I characteristics of Zener Diode
3. To Find V-I characteristics of Tunnel Diode
4. To Find V-I characteristics of Photo Diode
5. To find Input/output characteristics of common base PNP/NPN transistor.
6. To find Input/output characteristics of common emitter PNP/NPN transistor.
7. Determination of Energy band gap ( $E_g$ ) using PN Junction Diode.
8. Study of regulated power supply.
9. Determination of Energy band gap ' $E_g$ ' of Ge using Four Probe method.
10. To Study Frequency of Hartley oscillator
11. To Study Frequency of Wein bridge oscillator
12. Study of RC coupled amplifiers



# Shri Vaishnav Vidyapeeth Vishwavidyalaya

## B.Sc. (Hons)

### Choice Based Credit System (CBCS)(Batch 2019-2022)

SUBJECT CODE	Category	SUBJECT NAME	TEACHING & EVALUATION SCHEME								
			THEORY			PRACTICAL		Th	T	P	CR EDITS
			END SEM University Exam	Two Term Exam	Teachers Assessment*	END SEM University Exam	Teachers Assessment *				
BSHCH 305	HONS	ANALYTICAL CHEMISTRY & ADVANCED CONCEPTS OF GENERAL CHEMISTRY - I	60	20	20	0	0	4	0	0	4

Legends: L - Lecture; T - Tutorial/Teacher Guided Student Activity; P – Practical; C - Credit; Q/A – Quiz/Assignment/Attendance, MST Mid Sem Test.

**\*Teacher Assessment** shall be based on following components: Quiz/Assignment/Project/Participation in class, given that no component shall exceed more than 10 marks.

#### Course Objective:

- To develop the understanding of fundamentals of Analytical Chemistry and General Chemistry.
- To give basic knowledge of Analytical Chemistry.

#### Course Outcomes:

After completion of the course the students will be able to understand:

- Fundamentals of Chemistry.
- Fundamentals of Analytical Chemistry.

#### Analytical Chemistry & Advanced Concepts of General Chemistry - I

##### Unit I: Periodicity of Elements:

*s, p, d, f* block elements, the long form of periodic table. Detailed discussion of the following properties of the elements, with reference to *s* & *p*- block.

- Effective nuclear charge, shielding or screening effect, Slater rules, variation of effective nuclear charge in periodic table.
- Atomic radii (van der Waals)
- Ionic and crystal radii.
- Covalent radii (octahedral and tetrahedral )
- Ionization enthalpy, Successive ionization enthalpies and factors affecting ionization energy. Applications of ionization enthalpy.



# Shri Vaishnav Vidyapeeth Vishwavidyalaya

## B.Sc. (Hons)

### Choice Based Credit System (CBCS)(Batch 2019-2022)

- (f) Electron gain enthalpy, trends of electron gain enthalpy.  
(g) Electro negativity, Pauling's/ Mulliken's/ Allred Rachow's/ and Mulliken-Jaffe's Electro negativity scales. Variation of electro negativity with bond order, partial charge, hybridization, group electro negativity. Sanderson's electron density ratio.

#### Unit II : Chemistry of Aliphatic Hydrocarbons

##### A. Carbon-Carbon sigma bonds

Chemistry of alkanes: Formation of alkanes, Wurtz Reaction, Wurtz- Fittig Reactions, Free radical substitutions: Halogenation - relative reactivity and selectivity.

##### B. Carbon-Carbon pi bonds

Formation of alkenes and alkynes by elimination reactions, Mechanism of E1, E2, E1cb reactions. Saytzeff and Hofmann eliminations.

Reactions of alkenes: Electrophilic additions their mechanisms (Markownikoff/ Anti Markownikoff addition), mechanism of oxymercuration-demercuration, hydroboration-oxidation, ozonolysis, reduction (catalytic and chemical), syn and anti hydroxylation (oxidation). 1, 2- and 1, 4- addition reactions in conjugated dienes and, Diels-Alder reaction; Allylic and benzylic bromination and mechanism, e.g. propene, 1-butene, toluene, ethyl benzene.

Reactions of alkynes: Acidity, Electrophilic and Nucleophilic additions. Hydration to form carbonyl compounds, Alkylation of terminal alkynes.

#### Unit III :Solid State:

Nature of the solid state, law of constancy of interfacial angles, law of rational indices, Miller indices, elementary ideas of symmetry, symmetry elements and symmetry operations, qualitative idea of point and space groups, seven crystal systems and fourteen Bravais lattices; X-ray diffraction, Bragg's law, a simple account of rotating crystal method and powder pattern method. Analysis of powder diffraction patterns of NaCl, CsCl and KCl. Defects in crystals.

#### Unit IV: Chemical Kinetics

Order and molecularity of a reaction, rate laws in terms of the advancement of a reaction, differential and integrated form of rate expressions up to second order reactions, experimental methods of the determination of rate laws, kinetics of complex reactions (integrated rate expressions up to first order only): (i) Opposing reactions (ii) parallel reactions and (iii) consecutive reactions and their differential rate equations (steady-state approximation in reaction mechanisms) (iv) chain reactions. Temperature dependence of reaction rates; Arrhenius equation; activation energy. Collision theory of reaction rates, Lindemann mechanism, qualitative treatment of the theory of absolute reaction rates.

Surface chemistry: Physical adsorption, chemisorption, adsorption isotherms. nature of adsorbed state. Catalysis: Types of catalyst, specificity and selectivity.



# Shri Vaishnav Vidyapeeth Vishwavidyalaya

## B.Sc. (Hons)

### Choice Based Credit System (CBCS)(Batch 2019-2022)

#### Unit V : Optical methods of analysis:

Origin of spectra, interaction of radiation with matter, fundamental laws of spectroscopy and selection rules, validity of Beer-Lambert's law.

*UV-Visible Spectrometry:* Basic principles of instrumentation (choice of source, monochromator and detector) for single and double beam instrument; Basic principle of quantitative analysis: estimation of metal ions from aqueous solution, geometrical isomers, keto-enol tautomers. Determination of composition of metal complexes using Job's method of continuous variation and mole ratio method.

*Infrared Spectrometry:* Basic principles of instrumentation (choice of source, monochromator & detector) for single and double beam instrument; sampling techniques. Structural illustration through interpretation of data, Effect and importance of isotope substitution.

#### Books:

1. Lee, J.D. *Concise Inorganic Chemistry*, ELBS.
2. Douglas, B.E. and Mc Daniel, D.H., *Concepts & Models of Inorganic Chemistry*, Oxford.
3. Atkins, P. W. & Paula, J. de *Atkin's Physical Chemistry* 8th Ed., Oxford University Press.
4. Ball, D. W. *Physical Chemistry* Thomson Press, India.
5. Vogel, Arthur I: *A Test book of Quantitative Inorganic Analysis* (Rev. by GH Jeffery and others). The English Language Book Society of Longman.
6. Willard, Hobert H. *et. al: Instrumental Methods of Analysis*, Wardsworth Publishing Company, Belmont, California, USA.



**Shri Vaishnav Vidyapeeth Vishwavidyalaya**

**B.Sc. (Hons)**

**Choice Based Credit System (CBCS)(Batch 2019-2022)**